Teacher’s Note: You have two card sets to choose from for the chat stations.

The first set of five cards provides the type of reaction. Students must solve for the products.

The second set of five cards provides the entire equation. Students must determine what type of reaction each equation is.

Determine the product(s). Add responses to your handout.

Station 1: Single Replacement

**Fe + 2CuCl →**

Determine the product(s). Add responses to your handout.

Station 2: Decomposition

2CuO →

Determine the product(s). Add responses to your handout.

Station 3: Combustion

**CH4 + 2O2 →**

Determine the product(s). Add responses to your handout.

Station 4: Double Replacement

**NaCl + HBr →**

Determine the product(s). Add responses to your handout.

Station 5: Synthesis

**2H2 + O2 →**

Station 1

**Fe + 2CuCl → FeCl2 + 2Cu**

Station 2

2CuO → 2Cu + O2

Station 3

**CH4 + 2O2 → CO2 + 2H2O**

Station 4

**NaCl + HBr → NaBr + HCl**

Station 5

**2H2 + O2 → 2H2O**