TYPES OF CHEMICAL REACTIONS (TEACHER COPY)

Summarize each definition in your own words as you follow along with the video.

Chemical Reaction	Definition	Equation Set Up
Synthesis	A reaction in which two or more substances combine to form a single new substance.	A+B→AB
Decomposition	A reaction in which a compound breaks down into two or more simpler substances.	AB→A+B
Combustion	A reaction in which a substance reacts with oxygen gas, releasing energy in the form of light and heat.	$C_xH_x(O_2) + O_2 -> CO_2 + H_2O$
Single Replacement	A reaction in which one element replaces a similar element in a compound.	A+BC→AC+B
Double Replacement	A reaction in which the positive and negative ions of two ionic compounds exchange places to form two new compounds.	AB+CD→AD+CB