## **Making the Attraction Real**

Directions: Read each claim and determine from what you've learned whether the claim is correct or incorrect. Justify why each claim is correct or incorrect.

Claim: The stronger the attraction between the atoms or molecules, the more energy it will take to separate them. Correct / Incorrect (circle one) Justification: Claim: As a temporary dipole is established in one molecule, it induces a dipole in all surrounding molecules. Correct / Incorrect (circle one) Justification: \_\_\_\_\_ Claim: Polar molecules only have induced dipoles. Correct / Incorrect (circle one) Justification: Claim: Hydrogen bonds are stronger than dipole-dipole forces, despite not being a true bond. Correct / Incorrect (circle one) Justification: \_\_\_\_\_\_ Claim: London dispersion forces are present in all molecules and atoms. Correct / Incorrect (circle one) Justification: \_\_\_\_\_\_