1. Balanced Equation: 2KClO₃ → 2KCl + 3O₂

What is the mass of A given? \_\_\_\_\_\_\_\_\_\_\_\_

Which element/compound(B) are you solving for? \_\_\_\_\_\_\_\_\_\_\_\_\_\_

|  |  |  |  |
| --- | --- | --- | --- |
|   |   |  |   |
|   |   |   |   |

Answer: \_\_\_\_\_\_\_\_\_\_\_\_

2. Balanced Equation: 4NH₃ + 5O₂ → 4NO + 6H₂O

What is the mass of A given? \_\_\_\_\_\_\_\_\_\_\_\_

Which element/compound(B) are you solving for? \_\_\_\_\_\_\_\_\_\_\_\_\_\_

|  |  |  |  |
| --- | --- | --- | --- |
|   |   |  |   |
|   |   |   |   |

Answer:\_ \_\_\_\_\_\_\_\_\_\_\_

3. Balanced Equation: 2Al + 3Br₂ → 2AlBr₃

What is the mass of A given?\_ \_\_\_\_\_\_\_\_\_\_\_

Which element/compound(B) are you solving for? \_\_\_\_\_\_\_\_\_\_\_\_\_\_

|  |  |  |  |
| --- | --- | --- | --- |
|   |   |  |   |
|   |   |   |   |

Answer: \_\_\_\_\_\_\_\_\_\_\_\_