Density Cubes	Name:
Directions: Follow the directions for determining the density of water. Then, use the space	
below to document your observations for each cube you're done.	e. Answer the questions at the end when
Mass of graduated cylinder (g):	
Measure 10. mL of water into the graduated cylinder	er.
Mass of water and graduated cylinder (g):	
Mass of just water (g):	
Density of water $(1 \text{ g/mL} = 1 \text{ g/cm}^3)$:	

- 1. What is the same about all of the cubes?
- 2. What is different about the cubes?
- 3. What do those differences mean?
- 4. What is the main conclusion you draw from exploring mass and volume of the cubes?
- 5. What is the relationship between the density of water and the density of the cubes that sink and float?